GERMACRANOLIDES, GUAIANOLIDES AND EUDESMANOLIDES FROM GREENMANIELLA RESINOSA

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Abstract—The aerial parts of *Greenmaniella resinosa* afforded, in addition to known compounds, 22 new ones: nine germacranolides, seven eudesmanolides, five guaianolides and a bisabolene derivative. The structures were elucidated by spectroscopic methods and by some chemical transformations including partial synthesis. The chemotaxonomic situation of *Greenmaniella* and the biogenetic relationships of the compounds are discussed briefly.

INTRODUCTION

Greenmaniella Sharp. (Compositae, tribe Heliantheae), a monotypic genus confined to the Monterrey area of north-eastern Mexico, was originally described as a Zaluzania in the subtribe Helianthinae. It was placed later with some reservations in the subtribe Verbesininae [1] but subsequently transferred to the subtribe Neurolininae [2, 3]. As nothing is known on the chemistry of Greenmaniella G. resinosa we have investigated.

RESULTS AND DISCUSSION

The aerial parts of G. resinosa afforded as main constituents the germacranolides tagitinin C (3a) [4-6], tagitinin E (1b) [6], tagitinin F (5a) [4-6] and 1-desoxyorizabin (4a) [7] as well as the eudesmanolide pinnatifidin (9a) [8]. Furthermore, the germacranolides 1a, 2, 3b, 4b, 4c, 4d, 5b, 5c, 6 and 14, the eudsmanolides 7a, 7b, 7e, 8a, 8b, 9b and 9c, the guaianolides 10, 11, 12a, 12b and 13 as well as the known lactones ivalin [9], ovatifolin [10], desacetyltulipinolide- 1β , 10-epoxide [11] and 8-epidentatin A (7c) [12] and the bisabolen-diol 15 were isolated.

The structure of 1a followed from the ¹H NMR spectrum (Table 1) which was very similar to that of 1b. The presence of the corresponding propionate followed from the typical ABX₃ signals of a chiral propionate.

The ¹H NMR spectrum of 2 was obtained from a sample which was a mixture with a second lactone (Table 1). All efforts to separate this mixture were unsuccessful. It eventually emerged that we were dealing with an equilibrium mixture of two lactones which changed with temperature. Integration of the exomethylene proton signals indicated a 1:1.125 mixture at 25° and a 1.2:1 mixture at 77°. Spin decoupling allowed the assignment of all signals of both lactones. All data only agreed with the presence of ketone 2, which we have named resinosolide. The ¹H NMR data of the second lactone indicated the presence of the hemiacetal 4c. Treatment of the mixture

with a trace of acid in methanol afforded the acetal 4e; its ¹H NMR spectrum (Table 1) was close to that of 4c, but also to that of 4d, which differed from 4c only by the replacement of the 1-hydroxy by a methoxy group. Spin decoupling allowed the assignment of all signals of 4c-4e. The data of 4c indicated that this lactone was an isomer of orizabin [13]. In the spectrum of orizabin the H-1 signal is a slightly broadened doublet but in the spectra of 4c-4e this signal is replaced by a double doublet. The stereochemistry was determined by NOE difference spectroscopy. Clear effects were observed between H-14, H-8, H-1, OMe and H-2 α , between H-7 and H-8, between H-9 β and H-6 as well as between OMe, H-1 and H-9 β (always first proton saturated). Also, the 13C NMR data of 4d agreed with the proposed structure (Experimental). It is noteworthy that in the case of 4d no trace of the corresponding O-methyl derivative of 2 could be observed. Only one 1β hydroxy derivative has been reported so far from a Viguiera species [14]. The configuration of a second lactone from a Calea species (compound 7 in ref. [15]) has to be corrected to la-hydroxy.

A further contribution to the elucidation of the stereochemistry of these lactones is the result of the acid catalysed transformation of 1b to 4f. The ¹H NMR spectrum (Table 1) and inspection of a model indicated that in compound 4f the configuration at C-3 and C-10 was opposite to that of 4a-4e. Most likely 4f was formed by protonation of the epoxide followed by attack of the 3-hydroxy group which led to inversion at C-10. The ¹H NMR data are typically different from those of 4a-4e and related lactones.

The ¹H NMR spectrum of 3b (Table 1) was very close to that of 3a. The changed nature of the ester group followed from the typical signals of a methyl butyrate. The structure of 4b clearly followed from the ¹H NMR spectrum (Table 1) which was close to that of 4a though small differences led to overlapping of some signals. Addition of deuteriobenzene allowed the assignment of all signals by spin decoupling. Similarly, the ¹H NMR data of 5b and 5c (Table 1) were close to those of 5a indicating the

presence of the corresponding 3-O-methyl ether and the 8-O-methyl butyrate, respectively. As followed from the couplings of H-8 the conformation of 5a-5c differed from that of 4a-4c.

The ¹H NMR spectrum of 6 (Table 1) was in part similar to that of 3a. However, the pair of lowfield doublets (H-1 and H-2) was replaced by a pair of narrowly split doublets at δ 3.68 and 3.22 (J = 2.5 Hz). Furthermore the H-5 signal was shifted downfield (δ 6.36). Spin decoupling allowed the assignment of all the signals. Therefore, in agreement with the molecular formula, the presence of the epoxide 6 was proposed. A very similar compound with a 8-O-angelate group was prepared by oxidation of tifruticin, itself obtained by epoxidation of desoxyfruticin [5]. The ¹H NMR spectra, however, were clearly different. NOE difference spectroscopy with 6 indicated that only the configurations at C-1 and C-2 were different in these closely related lactones. Clear NOEs were observed between H-14 and H-2, between H-15 and H-5, between H-7, H-8 and H-1, between H-2, H-9 β , H-6 and H-14 as well as between H-6 and H-2. Inspection of a model indicated that these effects and the couplings agreed with the proposed stereochemistry and with a conformation where the ketone was nearly in plane with the conjugated double bond. This explained the downfield shift of the H-5 signal. In the case of dehydrotifruticin steric hindrance leads to a conformation where the keto group is not in plane with the double bond. Epoxidation of 3a afforded a single epoxide which is identical with the natural product. Inspection of a model indicated that in the case of 3a in its preferred conformation the attack of peracid should be favoured from one side by steric effects and by the 10-hydroxy group.

Obviously all these germacranolides are biogenetically related. Oxidation of 1b followed by hydrolysis of the epoxide would lead to 2 which by elimination of water would give a mixture of the E,Z-isomers 3a and 3c. Inspection of a model of 3c shows that it can be easily transformed into 5a and, accordingly, 3c was not observed. Epoxidation of 3a would give 6. The lactone 4a could be formed by hydrogenation of either 5a or 3a. The O-methyl ethers 4b and 5b may be artifacts as methanol was used for extraction. This also could be considered in the case of 4d as the corresponding 8-O-angelate was prepared from the corresponding dienone [5] by methanol addition.

The structure of 7a and 7e followed from the ¹H NMR spectra (Table 2) which were similar to those of reynosin [16] and balchanin [17], respectively. The presence of an 8β -isobutyryloxy group could be deduced from the typical signals of the isobutyrate residue and the couplings of H-8. The spectrum of 7c was close to that of 7a but, as expected, the H-8 signal was shifted upfield. Lactone 7c has been reported previously [12] but no NMR data were

1* 3Ь 46 **4c** 4e 45† 5b 5c 6 2.02 m H-1] 2.79 dd 4.19 dd 6.91 d 4.51 dd 4.02 dd 4.37 dd 3.92 d 5.68 d 5.81 d 3.22 d H-1' 1.80 m § 2.60 dd 2.49 dd 2.46 ddd 2.88 dd 2.52 dd H-2 2.22 m 2.03 ddd (6.23 d 6.27 d6.31 d 3.68 d H-2' 1.75 ddd 2.66 dd 2.11 dd 1.50 dd 2.04 dd 2.35 dd 2 02 m H-5 5.69 da 5.32 dq 5.73 dq 5.86 dq 5.66 da 5.64 da 5.61 dq 5.73 dq 5.40 ddg 5.68 da 6.36 da 5.90 ddq H-6 6.66 dd 5.30 dd 5.39 br d 5.34 ddq 5.43 ddq 5.39 ddq 5.38 ddq 6.43 br t 5.75 ddq 5.63 ddq H-7 2.84 dddd 3.23 dddd 3.53 br s 4.20 dddd 4.12 dddd 4.08 dddd 4.22 dddd 2.86 dddd 3.68 ddd 3.42 ddd 3.28 dddd H-8 5.18 ddd 5.40 ddd 5.37 ddd 5.59 dt 5.56 mt 5.62 ddd 5.64 dt 5.48 br d 5.22 dt 5.08 t 5.38 dd 1.97 dd H-9 2.73 dd 2.47 dd 1.84 dd 2.15 dd 1.73 dd 2.05 dd 2.34 dd 2.31 dd 2.36 dd 2.12 dd H-9' 1.32 dd 1.86 dd 1.98 dd 1.95 dd 1.78 dd 2.14 br dd 1.79 dd 1.53 br d 2.20 dd 2.28 dd 1.96 dd H-13 6.38 d 6.34 d 6.35 d 6.22 d 6.26 d 6.24 d 6.26 d 6.32 d 6.28 d6.30 d 6.35 d H-13' 5.76 d 5.76 d 5.80 d 5.57 d 5.60 d 5.57 d 5.61 d 5.69 d 5.59 d 5.65 d 5.84 d H-14 1.50 s 1.32 s 1.54 s 1.45 s 1.51 s 1.53 s 1.48 s 1.36 s 1.43 s 1.41 s 1.28 s H-15 1.82 d 2.00 br s 1.96 d 1.73 t 1.82 t 1.81 t 1.73 t 1.64 br s 1.88 t 1.93 t 1.95 t **OCOR** 2.33 ABX₃ 2.43 qq 2.42 qq 2.40 qq 2.41 qq 2.57 qq 2.48 qq 2.32 tq 2.49 qq 2.62 tq 2.38 qq 1.11 t 1.07 d 1.58 m 1.03 d 1.07 d 1.06 d 1.05 d 1.14 d 1.13 d 1.58 m 1.12 d 1.40 m 1.05 d 1.02 d 1.05 d 1.04 d 1.04 d 1.42 m 1.11 d 1.13 d 1.11 d 0.86 t 0.84 t 1.04 d 1.10 d OMe 3.20 s 3.38 s 3.10 s 3.29 s

Table 1. ¹H NMR spectral data of compounds 1, 2, 3h, 4h, 4c, 4d, 4e, 4f, 5h, 5c and 6 (CDCl₃, 400 MHz, δ values)

J (Hz): compound 1: 1,2 = 4.5; 1,2' = 10; 2,2' = 15; 2,3 = 4; 2',3 = 2.5; 5,6 = 11; 5,15 = 1.5; 6,7 = 2; 7,8 = 7,13 = 2.5; 7,13' = 2; 8,9 = 4.5; 8,9' = 2.5; 9,9' = 15; compound 2: 1,2 = 3.5; 1,2' = 7.5; 2,2' = 15; 5,6 = 8.5; 5,15 = 1.3; 6,7 = 7,8 = 2.5; 7,13 = 2; 7,13' = 1.8; 8,9 = 4.5; 8,9' = 10; 9,9' = 15; compound 3b: 1,2 = 16; 5,6 = 9; 5,15 = 1.5; 7,8 = 3; 7,13 = 2; 7,13' = 1.7; 8,9 = 6; 8,9' = 9; 9,9' = 14; compounds 4b-4e: 1,2 = 6; 1,2' = 10; 1,22' = 12.5; 1,23' = 10; 1,23' =

7a 7b 7c 7d 7e
R¹ *i*-Bu *i*-Bu H *i*-Bu *i*-Bu
R² H₂ β-OH, H H₂
$$\Longrightarrow$$
 O H₂
Δ3

8a
 8b
 8c

 R

$$CH_2$$
 CH_2
 CH_3 , H

 R
 4β -H
 4α -H
 4α -H

9a R =
$$\alpha$$
-H
9b R = α -OH

^{*}H-3 4.44 br dd.

[†]H-3 4.80 br dd.

[‡]In CDCl₃-C₆D₆ H-8 5.46 dt.

Table 2. ¹ H NMR spectral data of compounds 7a-7e, 8a-8c, 9b and 9c (CDCl ₃ , 400 MHz, δ val

	7a	7 b	7c	7 d	7e	8a*	8 b †	8c	9 b	9c
H-1	3.53 dd	3.57 dd	3.50 dd	3.90 ddd	3.68 dd	3.71 dd	3.76 dd	3.71 dd	2.64 br d 2.15 d	2.48 d 2.37 d
H-2 H-2'	1.83 <i>br ddd</i> 1.62 <i>ddd</i>	2.26 ddd 1.57 ddd	1.83 m 1.64 m	2.93 dd } 2.46 dd }	2.40 m	2.75 dd 2.59 m	2.68 m }	2.66 m		
H-3 H-3'	2.36 ddd \ 2.14 ddd \	4.11 br dd	2.35 ddd 2.14 ddd	_ }	5.37 br s			-	5.84 br s	5.87 br s
H-5 H-6	2.25 br d 4.51 t	2.13 br d 4.57 t	2.24 br d 4.62 t	2.61 dt 4.56 t	2.43 br d 4.42 t	1.66 dd 4.48 t	2.01 dd 4.51 t	1.89 dd 4.78 t	2.11 dd 1.63 dd	
H-7 H-8	2.83 dddd 5.76 ddd	2.82 dddd 5.75 ddd	2.72 dddd 4.66 br s	2.90 dddd 5.81 ddd	2.78 dddd 5.76 ddd	2.82 dddd 5.78 ddd	2.80 dddd 5,78 ddd	2.01 m 5.56 ddd	•	id 3.22 br dddi 4.74 ddd
H-9 H-9'	1.59 br d 2.32 dd	2.34 dd 1.58 dd	2.32 dd 1.54 dd	2.42 dd 1.68 dd	1.98 m 2.40 m	2.42 dd 1.58 br d	2.35 dd 1.59 br d	2.33 dd 1.62 dd	2.25 dd 2.00 br d	1.88 dd 1.81 dd
H-13 H-13'	6.16 d 5.44 d	6.17 d 5.46 d	6.35 d 5.53 d	6.22 d 5.52 d	6.16 d 5.43 d	6.19 d 5.47 d	6.19 d } 5.48 d }	1.29 d	6.18 d 5.65 d	6.36 d 5.76 d
H-14	0.99 s	0.97 s	1.07 s	1.11 s	1.07 s	1.30 s	1.28 s	1.27 s	1.13 s	1.09 s
H-15	{ 5.03 br s } { 4.96 br s }	5.36 br s 5.13 br s	5.02 br s 4.95 br s	{ 6.20 d { 5.84 d	1.89 br s	1.32 d	1.32 d	1.33 d	2.00 d	2.07 d
OCOR	2.54 qq 1.15 d 1.14 d	2.54 qq 1.15 d 1.14 d	<u> </u>	2.56 dd 1.17 d 1.16 d	2.54 qq 1.16 d 1.15 d	2.56 qq 1.17 d 1.16 d	2.57 qq 1.18 d 1.17 d	2.56 qq 1.21 d 1.20 d	***************************************	

^{*}H-4 2.59 m.

presented. Therefore we have added the relevant data in Table 1. The spectrum of 7e was similar to that of the corresponding angelate [18].

The ¹H NMR spectrum of 7b (Table 1) was very close to that of 7a. The presence of an additional hydroxy group at C-3 was deduced from the broadened double doublet at δ 4.11. The positions of the oxygen functions were established by spin decoupling and the configurations at C-1 and C-3 followed from the observed couplings. Those of H-3 differed clearly from those of trichomatolide B, a related 3α -hydroxy compound $\{19\}$.

The ¹H NMR spectra of 8a and 8b (Table 2) were similar indicating that these lactones may be epimers. This assumption was supported by spin decoupling which led to clear sequences for H-5 through H-9. Although some signals were overlapped the presence of a secondary methyl group at C-4 could be shown by decoupling. A double doublet at δ 3.71 and 3.76, respectively, indicated a secondary hydroxy group which could be placed at C-1. As the methyl singlets were shifted downfield (δ 1.30 and 1.28, respectively) in agreement with the molecular formula (C₁₉H₂₆O₆) the presence of guaianolides with a 1,10-epoxide was proposed. An alternative possibility was the presence of the eudesmanolides 8a and 8b. However, the chemical shift of H-14 would be unusual for this type of sesquiterpene lactone. As only minute amounts were isolated no reactions could be undertaken. However, a very weak 13CNMR spectrum was obtained which did not show signals for epoxide carbons. Thus the structures

of 8a and 8b were favoured. The ¹H NMR spectra in deuteriobenzene gave further information, especially the couplings of H-4, which clearly indicated that the epimer 8a had trans-diaxial protons at C-4 and C-5 while a 6 Hz coupling in the case of 8b required the cis-relationship of these protons. Accordingly, H-4 showed a W-coupling with H-2 which gave rise to an unresolved multiplet in the case of 8b. The presence of eudesmanolides were further supported by a W-coupling of H-14 with H-9α in both epimers. Finally the structure of 8b was established by a partial synthesis of 8c. Manganese dioxide oxidation of 7b afforded the ketone 7d which was stereospecifically hydrogenated to 8c. The ¹H NMR spectrum (Table 2) of the latter was nearly identical with that of 8b except for the signals of H-7, H-11 and H-13. Obviously the unusual shift of H-14 was due to deshielding effects of the 3-keto group and of the 8β -oxygen function. The fragmentation pattern in the mass spectra of 8a and 8b also supported the presence of eudesmanolides. In particular the fragment m/z 251 was probably formed by loss of ring A and this was followed by elimination of isobutyric acid leading to m/z 163 (C₁₀H₁₁O₂). Biogenetically 8a and 8b probably are formed from 1b which, by proton attack of the epoxide, could be transformed to an eudesmanolide cation. Loss of H-3 could lead then to the enois of 8a and 8b which are derivatives of artecalin [20]. However, the ¹H NMR data of the latter compound (8b) could not be compared with those of 8a as only a 60 MHz spectrum in deuterioacetone is reported in the literature [21].

[†]H-4 2.86 br dq.

J (Hz): compounds 7a and 7c: 1,2 = 4.5; 1,2' = 11; 2,2' = 2',3' = 3.3' = 12; 2,3 = 1.5; 5,6 = 6,7 = 11; 7,8 = 7,13' = 8.9 = 3; 7,13 = 3.5; 8,9' = 2; 9,9' = 15; compound 7b: 1,2 = 4.5; 1,2' = 2,2' = 2',3 = 12; 2,3 = 5; 5,6 = 6,7 = 11; 7,8 = 7,13' = 8.9 = 3; 7,13 = 3.5; 8,9' = 2; 9,9' = 15; compound 7d: 1,2 = 6; 1,2' = 10; 2,2' = 17; 5,6 = 6,7 = 10.5; 5,15' = 2; 7,8 = 3; 7,13 = 8.9' = 3.5; 7,13' = 3; 8,9 = 2.5; 9,9' = 15; compound 7e: 1,2 = 6; 1,2' = 9; H-5-H-9 and H-13 see 7a; compounds 8a and 8b: 2,2' = 15.5; 2,2' = 15.5; 2,2' = 10.5;

The structures of 9b and 9c followed from the ¹H NMR spectra (Table 2). All signals could be assigned by spin decoupling. While the spectrum of 9b was very close to that of pinnatifidin (9a) [8] that of 9c differed remarkably. The couplings of H-8 were changed and H-6 β was shifted downfield. The stereochemistry was determined by NOE difference spectroscopy which gave clear effects between H-14, H-9 α , H-9 β and H-2 β as well as between H-8, H-7 and H-2 α . Inspection of a model showed that these effects required the presence of a cis-decalin derivative where the B ring was in a boat conformation. Thus 9b and 9c were epimeric 5-hydroxypinnatifidins.

The ¹H NMR spectrum of 10 (Table 3) indicated the presence of a hydroperoxide ($\delta 8.45 \, s$) of a guaianolide where all signals could be assigned by spin decoupling. The stereochemistry followed from the couplings and from a strong NOE between H-14 and H-6 indicating a conformation with a quasi-axial 10-methyl group which agreed with the couplings of H-8. The presence of a hydroperoxide was supported by the mass spectrum. After loss of isobutyric acid (m/z 276) elimination of hydrogen peroxide was observed (m/z 242).

The ¹H NMR data of 11 (Table 3) showed that again a guaianolide was present. Most signals were close to those of dehydroleucodin derivatives [22]. However, as a signal for H-3 was missing in agreement with the molecular formula and the IR band at 3620 cm⁻¹ a 3-hydroxy derivative was present. A similar tiglate, but lacking the 1,10-double bond, was reported previously [23].

The ¹H NMR spectra of 12a and 12b (Table 3) indicated that epimers were most likely present again. Spin decoupling showed that a 3-ketoguaianolide with a 8β -isobutyryloxy group should be proposed. In agreement with the molecular formula the chemical shift of the methyl signals indicated hydroxy groups at C-4 and C-10. This was established by the 13C NMR spectrum of 12a (Experimental). The configurations at C-4 and C-10 were determined by NOE difference spectroscopy. In the case of 12a saturation of H-14 gave clear NOEs with H-9a, H- 9β and H- 2β while H-15 showed a NOE with H-5. Surprisingly in the ¹H NMR spectrum of 12b some signals were broadened (H-6 and H-7) which were sharpened at elevated temperature. Obviously the changed stereochemistry led to some flexibility of the conformation. The stereochemistry followed from the results of NOE difference spectroscopy. Clear effects were observed between H-14, H-9 β , H-6, H-2 β and H-1, between H-15 and H-6 and between H-7, H-5 and H-8. A weak NOE between H-14 and H-15 supported the proposed conformational flexibility.

The ¹H NMR spectrum of 13 (Table 3) indicated that an anhydro derivative of 12a/b was present. Accordingly, the signal of H-6 was a broadened doublet at δ 5.41 which was sharpened by irradiation at δ 2.00 (H-15) and 3.32 (H-1). All signals were assigned by spin decoupling and the stereochemistry was determined by the observed NOEs between H-14 and H-9 β as well as between H-7 and H-8.

The ¹H NMR spectrum of 14 (Table 3) was in part

Table 3. ¹H NMR spectral data of compounds 10, 11, 12a, 12b, 13 and 14 (CDCl₃, 400 MHz, δ values)

	10*	11	12a (C ₆ D ₆)	12b (60°)	13	14
 H-1	2.91 d	_	1.97 ddd	2.97 ddd	3.32 m	2.86 dd
H-2		_	2.11 dd	2.59 dd)	2.60	2.20 m
H-2'			1.80 <i>dd</i>	2.28 dd (2.60 m	1.56 m
Н-3	5.45 br s	_	_	_ ,	-	2.42 ddd 2.29 br dd
H-4	_	-	_	_	_	_
H-5	3.19 <i>ddq</i>	3.39 br d	1.91 t	2.71 dd	_	5.36 br d
H-6	4.67 dd	4.03 t	4.79 t	4.40 dd	5.41 br d	5.27 t
H-7	3.11 <i>dddd</i>	3.12 <i>dddd</i>	3.52 dddd	3.85 <i>dddd</i>	3.44 br ddd	2.77 dddd
H-8	5.71 ddd	5.68 dt	5.51 <i>ddd</i>	5.64 <i>ddd</i>	4.81 br dd	4.59 br d
H-9	2.29 dd	2.86 dd	2.15 dd	2.31 dd	2.37 dd	1.14 br d
H-9'	2.06 dd	2.75 br d	1.22 dd	1.82 dd	2.10 br dd	3.01 dd
H-13	6.34 d	6.24 d	6.25 d	6.27 d	6.37 d	6.37 d
H-13'	5.60 d	5.49 d	5.16 d	5.47 d	5.66 d	5.57 d
H-14	1.20 s	2.41 s	0.54 s	1.35 s	1.09 s	4.68 d 3.78 d
H-15	2.26 t	2.23 br s	1.48 s	1.52 s	2.00 dd	1.85 br s
OCOR	2.48 qq	2.48 qq	2.22 qq	2.48 qq	2.54 qq	2.13 s
	$1.11 \ \hat{d}$	1.10 d	1.01 d	1.11 d	1.15 d	
	1.09 d	1.09 d	0.96 d	1.09 d	1.14 d	

^{*}OOH 8.45 s.

J (Hz): compound 10: 1.5 = 6; 3.15 = 5.15 = 1; 5.6 = 11; 6.7 = 9.5; 7.8 = 2.5; 7.13 = 3.5; 7.13' = 3; 8.9 = 3; 8.9' = 5; 9.9' = 15; compound 11: 5.6 = 10; 7.8 = 1; 7.13 = 3; 7.13' = 2.5; 8.9 = 6; 8.9' = 1; 9.9' = 15; compound 12a: 1.2 = 5.5; 1.2' = 11; 1.5 = 5.6 = 6.7 = 9.5; 2.2' = 17; 7.8 = 7.13 = 3.5; 7.13' = 3; 8.9 = 8.5; 8.9' = 5; 9.9' = 15; compound 12b: 1.2 = 9; 1.2' = 1.5 = 7.5; 2.2' = 18.5; 5.6 = 11; 6.7 = 8.5; 7.8 = 3; 7.13 = 3.5; 7.13' = 3; 8.9 = 5; 8.9 = 9; 9.9' = 15; compound 13: 6.7 = 11; 1.15 = 6.15 = 1.5; 7.13 = 7.13' = 8.9' = 3; 8.9 = 3.5; 9.9' = 14.5; compound 14: 1.2 = 2; 1.2' = 11; 2.3 = 3.3' = 12; 2'.3 = 2.3' = 5; 5.6 = 6.7 = 10; $7.8 \sim 1$; 7.13 = 3.5; 7.13' = 3; 8.9 = 6.5; 9.9' = 15; 14.14' = 11.

similar to that of ovatifolin [10]. However, the replacement of the 1,10-double bond by an epoxide was indicated by an upfield shift of H-14 and the chemical shift of H-1 $(\delta 2.86, dd)$ which required the presence of an epoxide. The HNMR data were close to those of the corresponding 8-O-methacrylate [24].

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The structure of 15 could also be deduced from the ¹H NMR spectrum (Experimental). All signals were assigned by spin decoupling which led to the proposed sequence. The observed couplings and inspection of a model indicated the probable presence of a derivative of α zingiberene, a hydrocarbon also isolated from Calea species, which are probably related to Greenmaniella [25, 26]. The proposed configuration at C-5 agreed with the couplings and for biogenetic reasons a trans-diol was assumed.

non-polar fractions afforded heptadeca-1,7E,9E,15E-tetraen-11,13-diyne (centaur X₄) [27], geranylcymol [28] and caryophyllen epoxide.

The chemistry of Greenmaniella shows clear relationships to the genus Calea [29, 30]. The lactones like tagitinin E are biogenetically close to furanoheliangolides and therefore the chemistry of Neurolaena [31] and Brasilia [32], which are also placed in the subtribe Neurolininae [3], also indicates a relationship. However,

the constituents of Zaluzania [33], Schistocarpha [34] and Bebbia [34], which have been placed in this subtribe [2], clearly differ and therefore their placement in other subtribes [3] would be supported by the chemistry. It is clear that the removal of Greenmaniella resinosa from Zaluzania, where it was originally placed, is strongly supported by the chemistry. Further investigations of related genera may solve some of the remaining problems.

EXPERIMENTAL

The air-dried aerial parts (600 g, voucher Turner 15625) were extracted with MeOH-Et₂O-petrol (1:1:1) at room temp. The extract obtained was separated as reported previously [35] first by CC (silica gel) into four fractions (CC 1-CC 4) which were further separated by HPLC (RP 8, ca 100 bar; H 1: MeOH-H₂O, 7:3; H 2: MeOH-H₂O, 3:2; H 3: MeOH-H₂O, 11:9) and/or by repeated prep. TLC (silica gel, PF 254; P1: Et₂O-petrol, 3:1; P2: CHCl3-C6H6-Et2O, 1:1:1; P 3: CHCl3-MeOH, 30:1; P 4: petrol). Finally the following compounds were obtained (in parentheses condition of isolation and R_f or R_i values): CC 1: 35 mg geranylcymol (P 4, R_f 0.85), 20 mg squalene (P 4, R_f 0.8) and 20 mg centaur X4 (P 4, Rf 0.70). CC 2: 80 mg caryophyllenepoxide. CC 3: 3 mg 1a (H 2, R, 1.8 min, P 3, R, 0.62), 200 mg 1b (H 2, R, 3.1 min, P 3, R, 0.70), 15 mg 2, 4c (H 3, R, 4.1 min), 400 mg 3a (H 2, R, 3.0 min), 2 mg 3b (H 2, R, 5.5 min), 300 mg 4a (H 2, R, 7.9 min, P 1, R_f 0.60), 1.5 mg 4b (P 1, R_f 0.75), 150 mg 4d (H 1, R_t 4.4 min), 300 mg 5a (H 1, R_t 5.0 min, P 1, R_f 0.28), 3 mg 5b (H 1, R_t 4.0 min, P 1, R_f 0.60), 2 mg 5c (H 2, R_t 7.9 min, P 2, R_f 0.50), 8 mg 6 (H 2, R_t 3.1 min, P 3, R_f 0.50), 2 mg 7a (H 1, R_t 5.0 min, P 2, R_f 0.50), 2 mg 7e (H 1, R_t 5.7 min), 5 mg 9b (H 2, R_t 1.8 min, P 3, R_f 0.32), 3 mg 9c (H 2, R_t 1.2 min, P 3, R_f 0.50), 1 mg 10 (H 3, R_t 5.7 min), 2.5 mg 8a (H 3, R_t 6.9 min), 1 mg 8b (H 3, R_t 6.7 min, P 3, R_f 0.45), 2 mg 11 (P 3, R_t 6.70), 10 mg 12a (H 3, R_t 3.2 min, P 3, R_f 0.45), 2 mg 15 (H 1, R_t 2.8 min, P 2, R_f 0.60), 50 mg ovatifolin (H 2, R_t 4.7 min), 10 mg ivalin (H 2, R_t 3.2 min) and 1 mg desacetyl-tulipinolide-1 R_t 10x-epoxide. CC 4: 500 mg 9a (crystallization), 3 mg 7b (H 3, R_t 6.7 min, P 3, R_f 0.50), 1.5 mg 7c (H 3, R_t 2.9 min, P 3, R_f 0.50), 1.5 mg 12b (H 3, R_t 4.5 min, P 3, R_f 0.50).

Desacyltagitinin E-8-O-propionate (1a). Colourless crystals, mp 125°; $IR \nu_{max}^{CHCl_3} cm^{-1}$: 3600 (OH), 1765 (γ -lactone), 1730 (CO₂R); MS m/z (rel. int.): 262.121 $[M-RCO_2H]^+$ (2.5) (calc. for $C_{15}H_{18}O_4$: 262.121), 244 $[262-H_2O]^+$ (3), 57 $[C_2H_5CO]^+$ (100).

Transformation of 1b to 4f. Compound 1b (30 mg) in 3 ml C_6H_6 was refluxed for 30 min with 10 mg p-toluene sulphonic acid. After shaking with NaHCO₃ soln prep. TLC of the reaction products (CHCl₃-MeOH, 30:1) afforded 15 mg 4f, colourless crystals, mp 173°; IR $v_{max}^{CHCl_3}$ cm⁻¹: 3625 (OH), 1760 (γ-lactone), 1730 (CO₂R); MS m/z (rel. int.): 350.173 [M]⁺ (0.7) (calc. for $C_{19}H_{26}O_6$: 350.173), 262 [M-RCO₂H]⁺ (11), 244 [262-H₂O]⁺ (9), 226 [244-H₂O]⁺ (6), 211 [226-Me]⁺ (10), 71 [C₃H₇CO]⁺ (100); $[\alpha]_D^{24}$ - 141° (CHCl₃; α 0.79).

Resinosolide (2). Colourless gum; $IR V_{max}^{CCL_4}$ cm⁻¹: 3590 (OH), 1780 (γ-lactone), 1730 (CO₂R), 1700 (C=CC=O); MS m/z (rel. int.): 366.168 [M] + (0.15) (calc. for $C_{19}H_{26}O_7$: 366.168), 348 [M - H_2O] + (0.9), 330 [348 - H_2O] + (1), 278 [M - RCO_2H] + (6), 260 [278 - H_2O] + (5), 242 [260 - H_2O] + (2), 71 [C_3H_7CO] + (100). To 5 mg of the mixture in 2 ml MeOH was added 5 mg p-toluene solphonic acid. After standing for 12 hr at 20° usual work-up afforded 5 mg 4e; colourless crystals, mp 176°; MS m/z (rel. int.): 380.184 [M] + (2) (calc. for $C_{20}H_{28}O_7$: 380.184), 349 [M - OMe] + (2), 292 [M - RCO_2H] + (37), 71 [C_3H_7CO] + (100).

Desacyltagitinin C-8-O-[2-methylbutyrate] (3b). Colourless oil; $1R v_{max}^{CHCl_3} cm^{-1}$: 3610 (OH), 1765 (y-lactone), 1730 (CO₂R); MS m/z (rel. int.): 362.173 [M] $^+$ (0.1) (calc. for $C_{20}H_{26}O_6$: 362.173), 260 [M $-RCO_2H$] $^+$ (7), 242 [260 $-H_2O$] $^+$ (3), 85 [RCO] $^+$ (44), 57 [85 -CO] $^+$ (100).

Desoxytagitinin B-3-O-methyl ether (4b). Colourless oil; IR $v_{max}^{CCL_4}$ cm⁻¹: 1775 (γ-lactone), 1735 (CO₂R); MS m/z (rel. int.): 364.189 [M]⁺ (6) (calc. for C₂₀H₂₈O₆: 364.189), 332 [M - MeOH]⁺ (7), 276 [M - RCO₂H]⁺ (17), 261 [276 - Me]⁺ (18), 232 [276 - CO₂]⁺ (57), 71 [C₃H₇CO]⁺ (100).

 2β -Methoxydesoxytagitinin B (4d). Colourless oil; IR $\nu_{\rm max}^{\rm CCL}$ cm⁻¹: 3590 (OH), 1775 (γ-lactone), 1725 (CO₂R); MS m/z (rel. int.): 380.184 [M] * (0.7) (calc. for C₂₀H₂₈O₇: 380.184), 362 [M - H₂O] * (1.4), 330 [362 - MeOH] * (1), 292 [M - RCO₂H] * (11), 260 [292 - MeOH] * (9), 71 [C₃H₇CO] * (100); ¹³C NMR (C₆D₆, C-1-C-15): 86.5 d, 41.3 t, 103.6 s, 140.5 s, 129.2 d, 70.8 d, 57.9 d, 75.2 d, 35.7 t, 81.7 s, 137.3 s, 169.5 s, 121.9 t, 22.4 q, 27.1 q; OCOR: 175.6 s, 34.3 d, 18.7 q, 19.2 q; [α]₂^{2d*} - 124° (CHCl₃; c 1.3).

Tagitinin F-3-O-methyl ether (5b). Colourless oil; IR $v_{\text{max}}^{\text{CCL}}$ cm⁻¹: 1770 (y-lactone), 1735 (CO₂R); MS m/z (rel. int.): 362.173 [M]⁺ (2.2) (calc. for C₂₀H₂₆O₆: 362.173), 331 [M - OMe]⁺ (8), 274 [M - RCO₂H]⁺ (14), 259 [274 - Me]⁺ (12), 71 [C₃H₇CO]⁺ (100); $\begin{bmatrix} \alpha \end{bmatrix}_D^{24} - 84^\circ$ (CHCl₃; c 0.3).

Desacyltagitinin F-8-O-[2-methylbutyrate] (5c). Colourless oil; IR $\nu_{\max}^{\text{CHCl}_3}$ cm⁻¹: 3580 (OH), 1760 (y-lactone), 1730 (CO₂R); MS m/z (rel. int.): 362.173 [M]⁺ (0.6) (calc. for C₂₀H₂₆O₆: 362.173), 260 [M-RCO₂H]⁺ (12), 242 [260-H₂O]⁺ (6), 85

 $[C_4H_9CO]^+$ (51), 57 $[85-CO]^+$ (100).

1β,2α-Epoxytagitinin C (6). Colourless, amorphous material, mp 110°; IR $v_{\rm max}^{\rm CHCl_3}$ cm⁻¹: 3630 (OH), 1770 (y-lactone), 1735 (CO₂R), 1690 (C=CC=O); MS m/z (rel. int.): 364.152 [M]⁺ (0.1) (calc. for C₁₉H₂₄O₇: 364.152), 276 [M − RCO₂H]⁺ (1.5), 248 [276 − CO]⁺ (3), 233 [248 − Me]⁺ (6), 123 (82), 71 [C₃H₇CO]⁺ (100); ¹³C NMR (CDCl₃, C-1−C-15): 58.1 d, 65.5 d, 193.5 s, 137.1 s, 141.5 d, 75.1 d, 49.6 d, 73.0 d, 42.3 t, 70.0 s, 135.5 s, 168.8 s, 125.0 t, 26.0 q, 20.2 q; OCOR: 176.1 s, 34.1 d, 18.6 q (2 ×); [α]_D^{2d} −69° (CHCl₃; c 0.53). To 20 mg 1b in 1 ml CHCl₃ 20 mg m-chloroperbenzoic acid and 20 mg NaHCO₃ were added. After 16 hr HPLC (H3, R, 4.7 min) afforded 10 mg 6, identical with the natural compound and unreacted material.

8 β -Isobutyryloxyreynosin (7a). Colourless oil; IR $\nu_{\text{max}}^{\text{CCl}_4}$, cm⁻¹: 3600 (OH), 1775 (γ -lactone), 1735 (CO₂R); MS m/z (rel. int.); 334.178 [M]⁺ (1.6) (calc. for C₁₉H₂₆O₅: 334.178), 246 [M - RCO₂H]⁺ (22), 228 [246 - H₂O]⁺ (37), 71 [C₃H₇CO]⁺ (100).

8 β -Isobutyryloxyridentin B (7b). Colourless oil; IR $\nu_{\rm max}^{\rm CHCl_3}$ cm⁻¹: 3600 (OH), 1760 (y-lactone), 1730 (CO₂R); MS m/z (rec. int.); 350.173 [M]⁺ (8) (calc. for C₁₉H₂₆O₆: 350.173), 262 [M - RCO₂H]⁺ (2.5), 244 [262 - H₂O]⁺ (4.5), 229 [244 - Me]⁺ (2), 71 [C₃H₇CO]⁺ (100); [α]²⁰ - 2° (CHCl₃; c 0.45). Compound 7b (3 mg) in 3 ml MeOH was stirred with 30 mg MnO₂ for 1 hr at room temp. Prep. TLC (CHCl₃-MeOH, 50:1) gave 1.5 mg 7d (R_f 0.4) which was hydrogenated in MeOH with Pd/BaSO₄ (5%). The reaction product was purified by HPLC (H 3, R_1 7.8 min); colourless oil; MS m/z (rel. int.): 253 [M - C₃H₇O₂]⁺ (10), 165 [253 - RCO₂H]⁺ (62), 71 [RCO]⁺ (100). 8 β -Isobutyryloxybalchanin (7e). Colourless oil; IR $\nu_{\rm max}^{\rm CHCl_3}$ cm⁻¹: 3610 (OH), 1765 (y-lactone), 1730 (CO₂R); MS m/z (rel.

cm⁻¹: 3610 (OH), 1765 (y-lactone), 1730 (CO₂R); MS m/z (rel. int.): 334.178 [M]⁺ (8) (calc. for C₁₉H₂₆O₃: 334.178), 246 [M - RCO₂H]⁺ (32), 228 [246 - H₂O]⁺ (11), 217 [246 - CHO]⁺ (12), 71 [C₃H₇CO]⁺ (100).

8 β -Isobutyryloxyartecalin (8a). Colourless crystals, mp 228°; IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3580 (OH), 1770 (γ -lactone), 1730 (CO₂R); MS m/z (rel. int.): 350.173 [M]⁺ (0.7) (calc. for C₁₉H₂₆O₆: 350.173), 332 [M-H₂O]⁺ (0.5), 262 [M-RCO₂H]⁺ (5), 251 [M-C₅H₇O₂]⁺ (26), 163.075 [251-RCO₂H]⁺ (25), 145 [163-H₂O]⁺ (56), 117 [145-CO]⁺ (71), 71 [C₃H₇CO]⁺ (100); 13C NMR (C₆D₆, C-1-C-15): 76.4 d, 50.2 t, 205.8 s, 52.5 d, 44.1 d, 77.4 d, 46.3 d, 65.2 d, 40.6 t, 42.8 s, 134.9 s, 173.5 s, 118.2 t, 30.1 q, 13.7 q; OCOR: 175.5 s, 34.3 d, 19.2 q, 18.8 q; [α]_D^{24*} - 5° (CHCl₃; c 0.21).

 8β -Isobutyryloxy-4-epiartecalin (8b). Colourless gum; $1R v_{max}^{CHCI_3}$ cm⁻¹: 3580 (OH), 1770 (y-lactone), 1730 (CO₂R); MS m/z (rel. int.): 350.173 [M]⁺ (0.7) (calc. for C₁₉H₂₆O₆: 350.173), 332 (0.7), 251 (24), 244 [332 - RCO₂H]⁺ (5), 163 (82), 145 (60), 117 (82), 71 (100).

 5α -Hydroxypinnatifidin (9b). Colourless crystals, mp 201°; IR $\nu_{\rm CHC^{1}_{3}}^{\rm CHC^{1}_{3}}$ cm $^{-1}$: 3600 (OH), 1765 (γ -lactone), 1670 (C=CC=O); MS m/z (rel. int.); 262.121 [M] $^{+}$ (7) (calc. for C $_{15}$ H $_{18}$ O $_{4}$: 262.121), 247 [M - Me] $^{+}$ (3.5), 244 [M - H $_{2}$ O] $^{+}$ (3), 229 [244 - Me] $^{+}$ (2), 111 (71), 61 (100); [α] $_{2}^{\rm D4}$ * + 166° (CHCl $_{3}$; c 0.08).

5β-Hydroxypinnatifidin (9c). Colourless crystals, mp 175°; $\text{IR }_{\nu_{\text{CHC}}}^{\text{CHC}_3}$ cm⁻¹: 3630 (OH), 1765 (γ-lactone), 1670 (C=CC=O); MS m/z (rel. int.): 262.121 [M]⁺ (18) (calc. for $C_{15}H_{18}O_4$: 262.121), 244 (2), 98 (100); $[\alpha]_{\text{D}}^{24^\circ}$ + 51° (CHCl₃; c 0.39).

2-Oxo-10α-peroxi-8β-isobutyryloxyguaia-3, 11(13)-dien-12,6α-olide (10). Colourless gum; $\Pi v_{max}^{CHG_3}$ cm⁻¹: 3590 (OOH), 1770 (γ-lactone), 1730 (CO₂R); MS m/z (rel. int.): 364.152 [M]⁺ (1.4) (calc. for C₁₉H₂₄O₇: 364.152), 346 [M - H₂O]⁺ (1.4), 276 [M - RCO₂H]⁺ (5), 258 [276 - H₂O]⁺ (6), 242 [276 - H₂O₂]⁺ (4), 71 [C₃H₇CO]⁺ (100).

3-Hydroxy-8β-isobutyryloxydehydroleucodin (11). Colourless crystals, mp 124°; IR $\nu_{\rm max}^{\rm CHCl}$, cm⁻¹: 3620 (OH), 1760 (γ-lactone),

1720 (CO₂R); MS m/z (rel. int.): 346.142 [M]⁺ (42) (calc. for C₁₉H₂₂O₆: 346.142), 258 [M - RCO₂H]⁺ (54), 71 [C₃H₇CO]⁺ (100); [α]_D^{D+} + 71° (CHCl₃; c 0.07).

4β,10α-Dihydroxy-3-oxo-8β-isobutyryloxyguaia-11(13)-en-12, 6α-olide (12a). Colourless, amorphous material, mp 105°; IR $\nu_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$: 3600 (OH), 1765 (γ-lactone), 1735 (C=O, CO₂R); MS m/z (rel. int.): 366.168 [M]* (0.9) (calc. for C₁₉H₂₆O₇: 366.168), 348 [M - H₂O]* (0.5), 278 [M - RCO₂H]* (1.5), 260 [278 - H₂O]* (2.5), 242 [260 - H₂O]* (1.4), 177 (37), 109 (66), 71 (44), 69 (100); 13 C NMR (CDCl₃, C-1-C-15): 45.4 d, 47.1 t, 214.2 s, 77.3 s, 53.3 d, 76.5 d, 47.7 d, 65.4 d, 39.4 t, 71.3 s, 135.0 s, 171.8 s, 120.8 t, 32.1 q, 22.9 q; OCOR: 176.4 s, 34.0 d, 19.2 q, 18.8 q; [α]_D^{2δ*} + 40° (CHCl₃; c 0.38).

 4α , 10α -Dihydroxy-3-oxo-8 β -isobutyryloxyguaia-11(13)-en-12, 6α -olide (12b). Colourless, amorphous material, mp 115°; IR $\nu_{max}^{CHCI_3}$ cm⁻¹: 3605 (OH), 1760 (γ-lactone), 1735 (C=O, CO₂R); MS m/z (rel. int.): 366.168 [M]⁺ (0.1) (calc. for C₁₉H₂₆O₇: 366.168), 278 [M - RCO₂H]⁺ (1), 260 [278 - H₂O]⁺ (1), 71 [C₃H₇CO]⁺ (100).

3-Oxo- 10α -hydroxy- 8β -isobutyryloxyguaia-4,11(13)-dien-12, 6α -olide (13). Colourless gum: IR $v \overset{CCL}{\subset} cm^{-1}$: 3460 (OH), 1780 (y-lactone), 1735 (CO₂R), 1710 (C=CC=O); MS m/z (rel. int.): 348.157 [M]+ (5) (calc. for $C_{19}H_{24}O_6$: 348.157), 278 [M - O=C=CMe₂]+ (22), 260 [M - RCO₂H]+ (24), 242 [260 - H₂O]+ (7), 71 [C₃H₇CO]+ (100); $[\alpha]^{20}_{D}$ + 25° (CHCl₃; c 0.14). 1α _10 β -Epoxyovatifolin (14). Colourless crystals, mp 124°;

 $1\alpha,10\beta$ -Epoxyovatifolin (14). Colourless crystals, mp 124°; IR $\nu_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$: 3600 (OH), 1760 (y-lactone), 1740 (OAc); MS m/z (rel. int.): 262.121 [M - HOAc] $^+$ (15) (calc. for C₁₅H₁₈O₄: 262.121), 244 [262 - H₂O] $^+$ (8), 111 (96), 95 (100).

 4α , 5β -Dihydroxybisabola-2,10-diene (15). Colourless oil; IR $v_{max}^{CHCl_3}$ cm⁻¹: 3580, 3400 (OH); MS m/z (rel. int.): 238.193 [M] + (1.5) (calc. for $C_{15}H_{26}O_2$: 238.193), 220 [M - H_2O] + (4), 202 [220 - H_2O] + (2), 151 [220 - C_5H_9] + (17), 109 [C_8H_{13}] + (56), 69 [C_5H_9] + (100); ¹H NMR (CDCl₃): 2.31 (br q, H-1), 5.66 (dd, H-2), 5.62 (br dd, H-3), 3.79 (dd, H-5), 1.83 (ddd, H-6), 1.70 (ddd, H-6'), 1.59 (m, H-7), 1.40 (dddd, H-8), 1.23 (dddd, H-8'), 2.03 (br dddd, H-9), 1.95 (br dddd, H-9'), 5.10 (br t, H-10), 1.69 (br s, H-12), 1.61 (br s, H-13), 0.88 (d, H-14), 1.32 (s, H-15) [J (Hz): 1,2 = 2; 1,3 = 1.5; 1,6 = 1,6' = 1,7 = 6; 5,6 = 7; 5,6' = 2.5; 6,6' = 9,9' = 14; 8,9 = 8,9' = 8',9' = 9,10 = 9',10 = 7].

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